The Mg²⁺ Requirements for Rho Transcription Termination Factor: Catalysis and Bicyclomycin Inhibition[†]

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ABSTRACT: Kinetic studies document that the essential *Escherichia coli* transcription termination factor rho utilizes Mg^{2+} and ATP as a substrate and requires a second Mg^{2+} ion for maximum poly(C)-dependent ATP hydrolysis activity. The velocity curves show a classic nonessential Mg^{2+} activation pattern in which Mg^{2+} augments hydrolysis by 39% and gives a K_1' for MgATP of 9.5 μ M in the presence of excess Mg^{2+} and a K_1 for MgATP of 21.2 μ M under limiting Mg^{2+} concentrations. Bicyclomycin (1), a commercial antibiotic that inhibits rho, weakened Mg^{2+} binding at the nonessential site and disrupted the nonessential Mg^{2+} activation pathway for poly(C)-dependent ATP hydrolysis. The K_i values for 1 were 23 μ M and 35 μ M under excess and limiting Mg^{2+} conditions, respectively, while the $K_{Mg(app)}$ for nonessential Mg^{2+} increased with increasing 1 concentrations. These findings, when combined with reported mechanistic studies, provide an emerging picture of key catalytic and substrate binding sites that are necessary for rho function and that are proximal to the 1 binding site.

The essential Escherichia coli rho factor (1) terminates select transcripts at the ends of genes and operons (2) and at regulatory points before and between genes (3). Rho is an RNA-binding protein that belongs to a family of helicases with a homohexameric structure shaped like a toroid (4a). The monomers are identical, consisting of 419 amino acid residues divided into two structural domains: the primary RNA-binding domain (residues 1–151) and the ATP¹ hydrolysis domain (residues 167-419) (5). A solution structure of the N-terminal 130 residues of rho has been solved using NMR techniques (6), and a crystal structure is also available (7). Structural information of the intact protein is lacking since hexameric rho has resisted crystallization. However, rho factor shows considerable sequence and predicted secondary structure similarity to the β -subunit of F₁-ATP synthase (8), and models of rho based on this structure have been presented (8-12). We observed a nearly perfect alignment of the predicted secondary structure of rho after residue 160, which included specific residues involved in ATP binding and catalysis (11, 12).

Rho binds to specific cytidine-rich areas of newly synthesized RNA at the primary RNA-binding site and makes advancing contacts with the transcript at the secondary RNA-

binding site in a process fueled by nucleoside triphosphate hydrolysis (4). A tethered tracking mechanism that results in disruption of the RNA polymerase—transcription complex, transcription termination, and the release of RNA and rho has been proposed for rho translocation (4).

The macromolecular target for the commercial antibiotic bicyclomycin (1) is rho (13). Mechanistic studies indicate that bicyclomycin interferes with rho RNA tracking and prevents protein conformational changes necessary for ATP hydrolysis (14). Bicyclomycin is projected to bind to a protein cleft in rho that is positioned at the interface of two rho subunits and close to the ATP binding and RNA tracking sites (15).

The sequence similarity between rho and F_1 -ATP synthase prompts the question: do these proteins share an equivalent enzyme mechanism? F_1F_0 -ATP synthase catalyzes the conversion of ADP and P_i to ATP where ATP synthesis occurs in nucleotide binding sites located in the F_1 sector (16). Detachment of the membrane-embedded F_0 protein to give F_1 -ATP synthase reverses the catalytic process permitting ATP hydrolysis. F_1 -ATP synthase is characterized by three catalytic MgATP sites that sequentially hydrolyze ATP and three sites that are noncatalytic. MgATP is also the suggested

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¹ Abbreviations: ATP, adenosine 5'-triphosphate; DTT, dithiothreitol; EDTA, ethylenediaminetetraacetic acid; PEI-TLC, poly(ethylenimine) thin-layer chromatography; SDS-PAGE, sodium dodecyl sulfate—polyacrylamide gel electrophoresis; TE, Tris-HCl and EDTA; Tris, tris-(hydroxymethyl)aminomethane.

substrate for rho function (17). Without $\mathrm{Mg^{2^+}}$ there is no ATP hydrolysis and no transcription termination (17c, 18a). Little is known about the molecular pathway for rho ATP hydrolysis, and only a few reports have appeared on the role of $\mathrm{Mg^{2^+}}$ in rho-mediated processes (4a, 18).

We report herein the Mg²⁺ requirements for rho-mediated poly(C)-dependent ATP hydrolysis. We show that maximal ATPase activity requires a *second* Mg²⁺ ion and further demonstrate that bicyclomycin binding inhibits the catalytic role provided by this second Mg²⁺ ion.

MATERIALS AND METHODS

Materials. Bicyclomycin was provided as a gift from Fujisawa Pharmaceutical Co., Ltd., and was purified by three successive silica gel chromatographies using 20% methanol—chloroform as the eluant. [γ -³²P]ATP (6000 Ci/mmol) was purchased from Perkin-Elmer (Boston, MA), Bio-Spin 6 columns were from Bio-Rad (Hercules, CA), and PEI-TLC plates used for ATPase assays were obtained from J. T. Baker, Inc. (Phillipsburg, NJ). Poly(C) was from Sigma (St. Louis, MO) and was dissolved in 100 μ L of TE buffer and dialyzed against aqueous 1.0 M potassium phosphate, pH 7.0 (8 h, 2×, 4 °C) using Slide-A-Lyzer cassettes from Pierce (Rockford, IL). All other chemicals were of reagent grade.

Bacterial Strains and Plasmids. Wild-type rho from E. coli was purified as described by Mott (19) from strain AR120 containing plasmid p39-ASE (20). Rho purity was determined by SDS-PAGE, and protein concentration was measured according to the Lowry assay (21).

Trp t' RNA Preparation and Isolation. The DNA substrate was a truncated form of the trp operon (22) modified and placed in the pTRP5 plasmid as described (14).

Trp t' RNA was made using SalI-linearized pTRP5 plasmid and T7 RNA polymerase. The reaction (100 μ L) was carried out in transcription buffer (40 mM Tris·HCl, 15 mM MgCl₂, 5 mM DTT, and 0.5 mg/mL acetylated BSA) containing 1 mM ATP, GTP, and UTP, 24 µM CTP, 40 units of RNase inhibitor, and 25 μ g of DNA. The reaction was started with 1.4 μ g of T7 RNA polymerase and was incubated at 42 °C for 2 h, another 0.56 µg of enzyme was added, and the reaction was incubated for another 2 h. RNase-free DNase (1 μ L) was added, and the RNA was purified using an RNeasy kit (Qiagen, Inc., Santa Clarita, CA). EDTA was added to a final concentration of 40 mM, and the trp t' RNA was dialyzed against 100 mM potassium phosphate, pH 7.0 (6 h, 1×, 4 °C), and 50 mM Tris·HCl, pH 7.9 (8 h, 2×, 4 °C). Purity was determined by urea-PAGE, and the RNA concentration was determined by UV at A_{260} using an extinction coefficient of 4.5 μ M⁻¹ cm⁻¹.

Spin Column Purification. Rho (10 μ M) in an aqueous 40 mM Tris•HCl (pH 7.9) solution containing EDTA (20 mM) was incubated on ice (10 min). Aliquots (100 μ L) of the protein solution were applied to Bio-Spin 6 columns, which were pretreated (2×) with 500 μ L of an aqueous 40 mM Tris•HCl (pH 7.9) solution containing 50 mM KCl and then centrifuged (2000 rpm, 4 °C, 2 min). Measurements of the spin column-treated rho showed no loss of protein (Lowry) and a residual activity of 10% or less in Mg²⁺-free buffer [poly(C)-dependent ATPase assay].

Poly(C)-Dependent ATPase Assay (23). The ribonucleotide-stimulated ATPase activity of rho at 32 °C was assayed

by measuring the amount of 32 P-labeled inorganic phosphate hydrolyzed from ATP after separation on PEI-TLC plates (prerun with water and dried) using 0.75 M potassium phosphate (pH 3.5) as the mobile phase (24). Reactions were initiated by adding ATP (varying concentrations) and 0.5 μ Ci of [γ - 32 P]ATP to the solution containing 40 mM TristhCl (pH 7.9), 50 mM KCl, 100 nM poly(C), 40 nM rho (monomer) and MgCl $_2$ (various concentrations). The TLC plates were exposed to PhophorImager plates (Fuji and Molecular Dynamics) (3 h), scanned on a Storm 860 PC PhosphorImager, and analyzed using Molecular Dynamic's ImageQuant 5.0. The initial rates of the reactions were determined by plotting the amount of ATP hydrolyzed against time. Each reaction was performed in duplicate, and the results were averaged.

Trp t'-Dependent ATPase Assay. The *trp t'*-stimulated ATPase activity of rho at 45 °C was assayed using the preceding procedure. Reactions were initiated by the addition of ATP (varying concentrations) and 0.5 μ Ci of [γ - 32 P]ATP to a solution containing 40 mM Tris·HCl (pH 7.9), 50 mM KCl, 65 nM *trp t'* RNA, 50 nM rho (monomer), MgCl₂ (varying concentrations), and 10% (v/v) DMSO.

Kinetics of Poly(C)-Stimulated Rho Activation and Inhibition by Bicyclomycin. ATPase assays were carried out as described in the preceding section using a six-channel, multiwell procedure. The reactions (100 μ L) were run with 2, 4, 5, 8, 10, 16, 20, 32, 40, 64, 80, 128, 160, 256, 320, 512, 640, 1048, 1280, and 2560 μ M ATP and 4, 8, 16, 32, 64, 128, 256, and 2560 μ M MgCl₂. Each series was repeated in the presence of 15, 30, and 60 μ M bicyclomycin. Samples were preincubated at 32 °C for 2 min prior to ATP addition, and five 1.4 μ L aliquots were spotted onto PEI-TLC plates after 15 s and then at selected times (10–300 s) depending on the ATP concentrations.

Kinetics of trp t'-Stimulated Rho Activation. ATPase assays were carried out as described in the preceding section using a six-channel, multiwell procedure. The reactions (40 μ L) were run with 8, 16, 32, 64, 128, 160, 256, 320, 512, 640, 1024, and 2560 μ M ATP and 32, 64, and 256 μ M MgCl₂. Samples were preincubated at 45 °C for 2 min before ATP was added, and five 1.4 μ L aliquots were spotted onto PEITLC plates after 15 s and then at selected times (10–300 s) depending on the ATP concentrations.

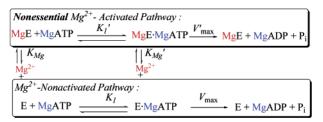
Isothermal Titration Calorimetry Measurements. The experiments were performed at 26 °C in a MicroCal (Northampton, MA) VP ultrasensitive isothermal titration calorimeter. An aqueous solution containing 50 mM Tris-HCl (pH 7.9) and 1 (20 mM) was added to a 1.4 mL sample cell. Aliquots (55 \times 5 μ L) containing 300 mM MgCl₂ in 50 mM Tris·HCl (pH 7.9) were injected over a 10 s period, and the heat of reaction (microcalories per second) over a 3.5 min interval was determined by integration of the peak areas using Origin software from MicroCal. Binding was complete within the first third of the titration, and the heat of dilution was determined from the baseline once titration was complete. The Origin software provided the best-fit values of ΔH° , the stoichiometry of binding (n), and the dissociation constant (K_d) from plots of heat evolved per mole of MgCl₂ injected versus the MgCl₂/1 molar ratio (24).

Table 1: Rho-Mediated Poly(C)-Dependent ATPase Activity at Near Constant MgATP Concentrations

[Mg _{tot}] (µM)	[ATP _{tot}] (µM)	[MgATP] ^a (µM)	$[\mathrm{Mg}_{\mathrm{free}}] \ (\mu \mathrm{M})$	$\begin{array}{c} [\text{ATP}_{\text{free}}] \\ (\mu\text{M}) \end{array}$	V^b (μ mol min ⁻¹ mg ⁻¹)
32	320	26.6	5.4	293.4	10.5 ± 0.01
80	80	34.5	45.5	45.5	12.7 ± 0.5
400	40	34.4	365.6	5.6	15.6 ± 1.6

^a Initial MgATP concentrations were calculated using a K_d (MgATP) = 60 μ M (28). ^b The values are the average of two determinations.

Chart 1: Kinetic Model of ATP Hydrolysis in Rho^a



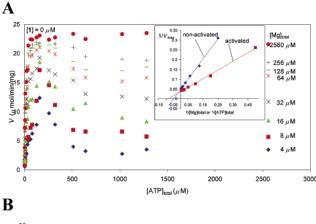
 $^{\it a}$ MgATP binds to the catalytic site, and free Mg $^{2+}$ is a nonessential activator.

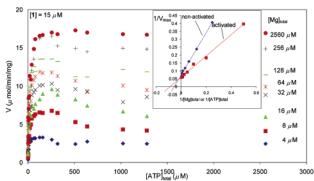
RESULTS

Rho protein was prepared as previously described (19) from the *E. coli* strain AR120 containing the overexpression plasmid p39-ASE and extensively dialyzed against buffer containing 0.1 mM EDTA. This preparation showed residual ATPase activity (23) of 35% in Mg²⁺-free buffer compared with 10 mM MgCl₂ buffer (25). When rho in ATPase buffer (25) containing EDTA (20 mM) was passed through a Bio-Spin 6 spin column (Bio-Rad, Hercules, CA), the activity was reduced to 10% of fully activated protein. However, when 10 mM Mg²⁺ was added, the ability to be 90% activated was regained. These findings indicated that a low level of Mg²⁺ or other ions bound tightly to rho.

Adding varying concentrations of Mg^{2+} and ATP (5'-ATP-Na₂) to the purified rho showed that ATP hydrolysis in the poly(C)-dependent assay did not depend solely on MgATP concentrations and that free Mg^{2+} augmented the conversion of ATP to ADP and P_i when the MgATP concentration was held at a near constant value (Table 1). This finding suggested that Mg^2 served as a nonessential activator (26) (Chart 1).

To document the nonessential Mg²⁺ activation pathway, we monitored the ATP hydrolysis at different Mg²⁺ concentrations (4, 8, 16, 32, 64, 128, 256, 2560 μ M) and at different ATP concentrations (2, 4, 5, 8, 10, 16, 20, 32, 40, 64, 80, 128, 160, 256, 320, 512, 640, 1048, 1280, 2560 μ M) (Figure 1A). The 160 reaction matrix was run in duplicate, and the average rates were plotted against ATP concentrations for the various Mg²⁺ concentrations (Figure 1A). The experimental design provided conditions under which the Mg²⁺ concentrations exceeded that of ATP, and the ATP concentrations were greater than Mg²⁺. Specifically, we observed a steady increase in the V_{max} for the reaction rates with increasing Mg²⁺ concentrations, an elevated hydrolysis rate at low ATP concentrations (128–320 μ M) for Mg²⁺ concentrations between 4 and 256 µM that decreased at higher ATP concentrations (512-2560 μ M), and the appearance of near hyperbolic velocity hydrolysis curves with high Mg²⁺ concentrations (2560 μ M). Figure 1A inset shows





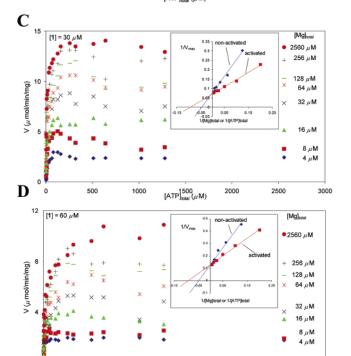


FIGURE 1: Velocity curves in poly(C)-dependent ATPase assay of rho vs total ATP concentration at various fixed Mg²⁺ concentrations. The reactions were conducted using a solution (100 μ L) containing ATPase buffer, rho [40 nM (monomer)], poly(C) (100 nM), and various concentrations of ATP and Mg²⁺ at 32 °C. The average velocities of two determinations are plotted. Inset: double reciprocal plot of the poly(C)-dependent ATPase activity of rho protein with either varying ATP (Mg²⁺ activated) or Mg²⁺ (Mg²⁺ nonactivated) concentrations. Panels: (A) $\mathbf{1} = 0~\mu$ M, (B) $\mathbf{1} = 15~\mu$ M, (C) $\mathbf{1} = 30~\mu$ M, and (D) $\mathbf{1} = 60~\mu$ M.

2500

500

Table 2: Key Thermodynamic and Kinetic Parameters for Poly(C)-Dependent ATP Hydrolysis in the Presence of 1

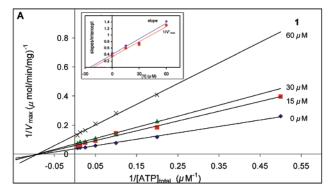
[1] (µM)	<i>K</i> ₁ (μΜ)	<i>K</i> ₁ ' (μΜ)	$K_{ m Mg(app)} \ (\mu m M)$	$V_{ ext{max(app)}^a}$ $(\mu ext{mol}$ $ ext{min}^{-1} ext{mg}^{-1})$	$V'_{ ext{max(app)}}^a$ $(\mu ext{mol}$ $ ext{min}^{-1} ext{mg}^{-1})$	$V_{ m max}/$ $V'_{ m max}$
0	21.2	9.5	12.9	20.0	27.8	0.72
15	19.4	11.4	17.5	13.5	16.9	0.80
30	22.2	10.7	95.2	12.8	14.1	0.91
60	20.3	10.8	135	7.7	7.8	0.99

^a The values are the average of two determinations.

the double reciprocal plot of 1/(rate of ATPase activity) versus 1/[ATP]_{total} for reactions performed in the presence of excess Mg^{2+} (Mg^{2+} activated; [Mg^{2+}] = 256 and 2560 μ M, [ATP] = 2-256 μ M) and a similar plot of 1/(rate of ATPase activity) versus 1/[Mg²⁺]_{total} for reactions run at high ATP concentrations (Mg²⁺ nonactivated; [Mg²⁺] = 4-256 μ M, [ATP] = 1280 and 2560 μ M) (26a). Extrapolation of the data points from the double reciprocal plot to the y-axis provided the V'_{max} (Mg²⁺ activated) and V_{max} (Mg²⁺ nonactivated). Correspondingly, when the data points were extended to the x-axis, we obtained the Briggs-Haldane constants K_1' (Mg²⁺ activated) and K_1 (Mg²⁺ nonactivated). We found that, in the presence of excess Mg^{2+} , the K_1' for MgATP was 9.5 μ M (27) and the V'_{max} was 27.8 μ mol min⁻¹ mg⁻¹ and that, under limiting Mg²⁺ conditions, the corresponding constant K_1 for MgATP was 21.2 μ M and the V_{max} was 20.0 μ mol min⁻¹ mg⁻¹ (Chart 1, Table 2). These findings indicate that, under conditions when excess Mg2+ exists in solution [$K_d(MgATP) = 60 \mu M (28)$], Mg^{2+} enhanced the ATP hydrolysis rate by 39%. The dissociation constant for Mg^{2+} binding to the rho nonessential Mg^{2+} binding site, K_{Mg} , was calculated using the rate equation (eq 1) corresponding to the kinetic scheme (Chart 1) (26a). We determined that $K_{\rm Mg}$ (12.9 μ M) was comparable with K_1' and K_1 (9.5–21.2 μ M) in the absence of **1**.

The experimental data in Figure 1A were simulated using eq 1. We employed a nonlinear regression program (SigmaPlot 2001), and the parameters ($V_{\rm max}$, $V'_{\rm max}$, K_1 , K_1' , $K_{\rm Mg(app)}$, K_0) were set to the graphically determined values, allowing a 15% variation except for the total metal concentration. Satisfactory plots were obtained for Mg²⁺ concentrations from 16 to 250 μ M (Supporting Information). At Mg²⁺ concentrations lower than 16 μ M the calculated velocities were lower than the experimentally observed values at ATP concentrations less than 256 μ M, which suggests that trace levels of metals in rho contributed to the observed ATP hydrolysis rates.

When we repeated this experiment at different **1** concentrations (15, 30, 60 μ M), we saw a steady decrease in the ATP hydrolysis rate and a *rapid* loss of the elevated rates for ATP hydrolysis at low-to-moderate Mg²⁺ concentrations (4–256 μ M) (Figure 1B–D). Panels A and B of Figure 2 show the double reciprocal plots for the 0–60 μ M **1** reactions in the presence of excess Mg²⁺ (Mg²⁺ activated) and excess ATP (Mg²⁺ nonactivated), respectively. Table 2 summarizes the thermodynamic and kinetic parameters determined from graphic analyses and shows that while the K_1 ' and K_1 values for MgATP remain constant as the **1** concentrations reach its I_{50} value (60–70 μ M) (I_3), $V'_{\text{max(app)}}$ approached $V_{\text{max(app)}}$. Significantly, when increasing amounts of **1** (0, 15, 30, 60 μ M) were added, the apparent dissociation constant for



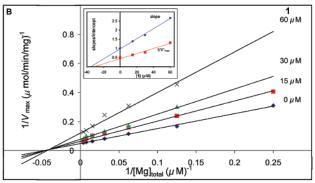


FIGURE 2: Effect of 1 on rho-mediated poly(C)-dependent ATPase activity: (A) excess Mg²⁺ employed; (B) excess ATP employed. See legend to Figure 1 for reaction conditions. Insets: plots of the slopes and intercepts versus 1 concentration.

nonessential Mg²⁺ binding to rho, $K_{\rm Mg(app)}$, progressively increased from 12.9 to 135 μ M (Table 2), indicating that the antibiotic weakened the protein interaction for the second, nonessential Mg²⁺ site. When the intercepts and slopes obtained in Figure 2 were plotted against 1 concentrations (Figure 2 insets), a linear response resulted, providing 1 K_i values of 23 and 35 μ M under excess and limiting Mg²⁺ conditions, respectively.

$$V = \frac{\frac{[\text{MgATP}]}{K_{1}} V_{\text{max}} + \frac{[\text{Mg}][\text{MgATP}]}{K_{\text{Mg}} K_{1}'} V'_{\text{max}}}{1 + \frac{[\text{Mg}]}{K_{\text{Mg}}} + \frac{[\text{MgATP}]}{K_{1}} + \frac{[\text{Mg}][\text{MgATP}]}{K_{\text{Mg}} K_{1}'}}$$
(1)

or in terms of [Mg]total and [ATP]total replace

$$\begin{split} [MgATP] &= \frac{[Mg]_{\text{total}} + [ATP]_{\text{total}} + K_o}{2} - \\ &\left\{ \left(\frac{[Mg]_{\text{total}} + [ATP]_{\text{total}} + K_o}{2} \right)^2 - [Mg]_{\text{total}} [ATP]_{\text{total}} \right\}^{1/2} \\ [Mg] &= [Mg]_{\text{total}} - \frac{[Mg]_{\text{total}} + [ATP]_{\text{total}} + K_o}{2} - \\ &\left\{ \left(\frac{[Mg]_{\text{total}} + [ATP]_{\text{total}} + K_o}{2} \right)^2 - [Mg]_{\text{total}} [ATP]_{\text{total}} \right\}^{1/2} \end{split}$$

where $K_0 = K_d(MgATP)$.

We examined three aspects of the proposed ATP hydrolysis pathway (Chart 1). First, we determined if the observed loss in rho poly(C)-dependent ATPase activity with increasing levels of ATP may have resulted from MgATP displacement by ATP. Rho solutions were prepared in which the

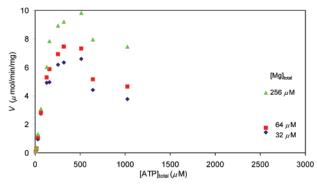


FIGURE 3: Velocity curves of rho in $trp\ t'$ -dependent ATPase assay vs total ATP concentration at various fixed Mg^{2+} concentrations. The reactions were conducted using a solution (40 μ L) containing ATPase buffer, rho [50 nM (monomer)], and $trp\ t'$ (65 nM) at various concentrations of ATP and Mg^{2+} at 45 °C in 10% (v/v) DMSO.

effective MgATP concentration was held at 400 μ M and the ATP concentration ranged from 400 to 4000 μ M. We observed no significant difference in rho poly(C)-dependent activity despite the 10-fold change in ATP concentration (data not shown). This finding was in agreement with an earlier report by Stitt that ATP did not bind to rho in either the presence or absence of poly(C) (17c).

We then asked if the nonessential Mg²⁺-activated pattern for rho-dependent ATP hydrolysis with poly(C) (Figure 1A) was observed for physiologically relevant RNA. Platt and co-workers have reported differences in the Mg²⁺ dependence of ATPase activation with poly(C) compared with natural RNA cofactors (18c). We chose $trp \ t'$, RNA previously shown to have a rho-dependent termination site (22), and conducted an abbreviated experiment corresponding to Figure 1A because of the limited quantities of this RNA. We determined the rates of ATP hydrolysis when Mg²⁺ was held at 32, 64, and 256 μ M and ATP ranged from 0 to 2560 μ M (Figure 3). We found that using our normal conditions [10 mM MgCl₂, aqueous Tris·HCl buffer (pH 7.9), 32 °C] trp t' afforded ATPase hydrolysis rates less than 5% of those observed with poly(C). When we ran the $trp\ t'$ experiments in aqueous Tris·HCl buffered solutions (pH 7.9) containing 0.5 mM MgCl₂ and 10% DMSO (v/v) and at 45 °C, we obtained ATP hydrolysis rates that were approximately 65% of those obtained for poly(C)-dependent processes run under the same conditions (data not shown). Richardson and Macy have shown that DMSO and higher temperatures destabilize the secondary structure of RNA permitting rho binding and enhancing ATP hydrolysis (29). Using these conditions for the trp t' RNA, we observed the hallmarks of nonessential Mg^{2+} activation kinetics: high V_{max} under excess Mg^{2+} conditions and low ATP concentrations (Mg²⁺ activated) and significantly lower $V_{\rm max}$ values when Mg²⁺ was limiting and the ATP concentrations were high (Mg^{2+} nonactivated).

Finally, we determined if the observed increase in the dissociation constant $K_{\rm Mg(app)}$ for the rho-Mg²⁺ complex when bicyclomycin was added stemmed, in part, from Mg²⁺ binding to **1**. Accordingly, we measured the binding isotherms after MgCl₂ addition (0–3 equiv) to aqueous 50 mM Tris+HCl (pH 7.9) buffered solutions containing **1**. We measured low levels of reaction heat when 1 mM **1** solutions were used. Elevating the **1** concentration to 20 mM led to detectable levels of heat, and best-fit analysis gave a binding

isotherm for a 1:1 Mg·1 complex with a $K_d = 5$ mM. Thus, the putative Mg·1 complex is ~80-fold weaker than the MgATP complex ($K_d = 60 \,\mu\text{M}$) (28). Similarly, we estimate that the Mg·1 species is ~400-fold weaker than the Mg·rho complex ($K_{\text{Mg(app)}} = 12.9 \,\mu\text{M}$; Table 2).

DISCUSSION

Previous studies showed that rho retained significant poly-(C)-dependent ATP hydrolysis activity without MgCl₂ and that hydrolysis is prevented by adding EDTA to the reaction solution (17c, 18a). We demonstrated that the use of spin columns and EDTA-pretreated rho provided a rapid method for removing low levels of divalent metal ions from purified rho solutions. The rho protein showed residual poly(C)dependent ATPase activity (10%), indicating that trace levels of divalent metals remained. Repetition of the spin column procedure did not reduce the level of ATPase activity. These findings were reminiscent of the earlier report by Lowery and Richardson showing that extensive dialysis of rho against EDTA-containing buffers still retained residual ATPase activity (18a) and studies by Senior and co-workers demonstrating that purified bovine F₁-ATP synthase retained a tightly bound Mg^{2+} (estimated $K_d \le 10^{-11} M$) (30).

Our kinetic studies revealed that a second Mg^{2+} facilitated ATP conversion to ADP and P_i and that it was necessary for maximal activity in the poly(C)-dependent assay (Figure 1A). The nonessential Mg^{2+} -activated pathway (Chart 1) accelerated ATP hydrolysis by 39%. Enhanced rho-dependent ATPase activity is likely to increase transcription termination efficiencies at prescribed RNA sequences. Indeed, we observed nonessential Mg^{2+} kinetic activation of ATP hydrolysis with $trp\ t'$ RNA (22) (Figure 3).

How does the nonessential Mg²⁺ catalyze ATP hydrolysis? There are several possibilities. The nonessential Mg²⁺ may facilitate ATP cleavage to ADP and Pi, enhance MgATP binding, chaperon MgATP to the nucleotide binding site, and assist in ADP removal. The kinetic data are compatible with each of these pathways. Clues concerning the role of the nonessential Mg²⁺ come from the inhibitory effect of 1 on this process (Figure 1B-D). We observed that addition of 1 led to the complete loss of the nonessential Mg²⁺ activation and a 10-fold increase in $K_{\text{Mg(app)}}$ from 12.9 to 135 μM when the bicyclomycin concentration reached its I_{50} value. This finding indicated that 60 μ M 1 fully prevented the nonessential Mg²⁺-activated process and reduced the V'_{max} to 28% of its former value (V'_{max} without $\mathbf{1} = 27.8 \ \mu\text{mol}$ $min^{-1} mg^{-1}$; V'_{max} with 60 μ M **1** = 7.8 μ mol min⁻¹ mg⁻¹). The pronounced loss of the nonessential Mg²⁺-activated ATP hydrolysis pathway upon 1 rho binding suggests that 1 either weakens ATP (MgATP) binding or perturbs the nonessential Mg²⁺-assisted ATP hydrolysis pathway or both, rather than disrupting nucleotide transport.

Graphic analysis of the data provided K_i values for 1 of 23 μ M and 35 μ M under saturating and limiting Mg²⁺ conditions, respectively, and showed that the nonessential Mg²⁺-activated pathway (saturating Mg²⁺ levels) was uniquely sensitive to added 1. Disruption of the nonessential Mg²⁺ activation process led to an appreciable loss of rho-mediated poly(C)-dependent ATPase activity, and thus less 1 (23 μ M) was required to inhibit total rho activity by one-half. Nonetheless, ATPase activity remained under conditions in

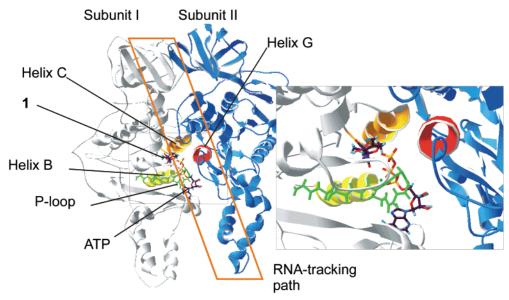


FIGURE 4: A model of the bicyclomycin binding pocket. The key helices [B (yellow) and C (brown)] for antibiotic binding, the ATP binding site, the P-loop (green backbone), and helix G (red) on the neighboring subunit are shown. The projected secondary RNA tracking site facing the central hole of rho is given in the box. Inset: a blow-up of the 1 binding pocket showing the proximity of the ATP hydrolysis site and helix G. The model was created using Swiss pdb Viewer (31).

which the nonessential Mg^{2+} -activated pathway does not occur (e.g., $1 = 60 \,\mu\text{M}$, Figure 1D, $V_{\text{max}} = 7.7 \,\mu\text{mol min}^{-1}$ mg⁻¹), and use of higher levels of 1 (e.g., $1 = 400 \,\mu\text{M}$) led to the near total loss of poly(C)-dependent ATPase activity (27b). This finding indicates that the antibiotic disrupts several catalytic pathways necessary for protein function. In agreement with these results, we have shown that 1 inhibits RNA binding to the rho secondary site leading to the loss of ATPase activity (14).

We, and others, have reported a structural model for rho based on the sequence and predicted secondary structural similarity to the β -subunit of F₁-ATP synthase (8–12). We have positioned the bicyclomycin binding pocket to be between the rho B and C helices with the 1 C(1) triol group pointing toward the outside of rho and the C(5)-C(5a)exomethylene unit pointing toward the terminal γ -phosphate and the P-loop (residues 175-183) (Figure 4) (15). Positioning 1 at this site places the antibiotic at or near the catalytic ATP site, but not at the ATP binding site, and within a few angstroms of the G-helix on the neighboring subunit (residues 332-336) that forms part of the extended secondary RNA tracking site (32, 33). A similar projection of the bicyclomycin binding pocket has been presented by Richardson and co-workers (34). Our findings that 1 inhibits the nonessential Mg²⁺-activated pathway contribute to an emerging picture of rho (12) showing that the key catalytic and substrate binding sites (12, 15, 33, 35) necessary for rho function are close to the putative 1 binding site (12, 15, 34). Current studies focus on determining the specific rho sites for Mg²⁺ binding and learning how Mg²⁺ assists ATP hydrolysis.

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SUPPORTING INFORMATION AVAILABLE

Simulated kinetic profiles for the poly(C)-dependent kinetics for reactions using $16-256\,\mu\text{M}$ MgCl₂ and $2-2560\,\mu\text{M}$ ATP. This information is available free of charge via the Internet at http://pubs.acs.org.

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